### SAFETY DATA SHEET



### **HYDROFLUORIC ACID**

# SECTION 1: Identification of the substance/mixture and of the company/undertaking Trade name:

1.1 Product identifier:	
CAS Number:	7664-39-3
EC number:	231-634-8
1.2 SYNONYMS:	<ul><li>Hydrogen fluoride solution</li><li>Fluorhydric acid</li></ul>
	<ul><li>HF (chemical formula)</li><li>Fluoric acid</li></ul>

### **SECTION 2: Hazards identification:**

2.1 Classification of the substance or mixture:	Classification according to Regulation (EC) No 1272/2008 The substance is classified according to the CLP regulation.
2.2 Label elements:	Labelling according to Regulation (EC) No 1272/2008 Acute toxicity, oral (category 3) Acute toxicity, dermal (category 3) Skin corrosion/damage (category 1A) Serious Eye Damage/Eye Irritation (Category 1) Specific Target Organ Toxicity (Single Exposure) (Category 1) Environmental Hazards (Aquatic Toxicity)(category 1)
Hazard Pictograms:	
Signal Word:	Danger
Hazard statements:	H301: Toxic if swallowed. H311: Toxic in contact with skin. H331: Toxic if inhaled. H314: Causes severe skin burns and eye damage. H318: Causes serious eye damage. H400: Very toxic to aquatic life. H410: Very toxic to aquatic life with long-lasting effects.



	H370: Causes damage to organs.
Precautionary Statements:	<b>P260:</b> Inhalation of Hydrofluoric
	acid fumes, vapors, or mist is
	hazardous; proper ventilation and
	respiratory protection are crucial.
FOTB	P262: Prevent skin and eye contact
	with Hydrofluoric acid, as it can
	cause severe damage.
	<b>P280:</b> Always wear appropriate
	personal protective equipment
	(PPE), including gloves, eye
	protection, and protective clothing,
	when handling Hydrofluoric acid.
	P301+P312: In case of ingestion,
	immediate medical attention is
	required. Poisoning can occur if
	swallowed.
	P302+P352: If Hydrofluoric acid
	comes into contact with the skin, it
	should be washed off immediately
	with soap and water.
	<b>P303+P361+P353:</b> If the acid is on
	skin or clothing, remove
	contaminated clothing and rinse
	skin thoroughly with water.
	<b>P305+P351+P338:</b> In case of eye contact, rinse with water and
	remove contact lenses if necessary.
	Seek medical help immediately.
	<b>P310:</b> If exposure occurs, immediate
	medical attention is necessary due
	to the severe hazards.
VOUD OUT MIC	<b>P405:</b> Hydrofluoric acid should be
YOUR CHEMIC	stored in a secure location to
1 5 5 11 5 11 E WITG	prevent unauthorized access,
	especially given its corrosive and
	toxic properties.
	<b>P501:</b> Proper disposal is essential to
	avoid environmental
	contamination, particularly to water
	bodies, due to Hydrofluoric acid's
	toxicity to aquatic life.
2.3 Other hazards:	



Inhalation:	can cause severe respiratory irritation, pulmonary edema, and systemic toxicity, potentially leading to death.
Ingestion:  ESTD	can cause severe gastrointestinal burns, systemic toxicity, and damage to internal organs such as the liver, kidneys, and bones, potentially leading to death.
Skin Contact:	can cause severe chemical burns, deep tissue damage, and systemic toxicity, potentially affecting bones, kidneys, and other organs if not treated promptly.
Eye contact:	can cause severe eye damage, including permanent blindness, corneal burns, and deep tissue injury.
Chronic Exposure:	At low concentrations, it can lead to cumulative systemic toxicity, affecting bones, teeth, kidneys, and the liver, potentially causing fluorosis, renal damage, and other long-term health issues.
Aggravation of pre-existing conditions:	may aggravate pre-existing respiratory conditions (like asthma or COPD) and skin disorders (such as dermatitis or eczema)

## **SECTION 3: Composition/information on ingredients**

3.1 Chemical characterisation:	Substances
CAS No:	Description: 7664-39-3 hydrofluoric acid
Identification number(s):	EC number: 231-634-8

#### **SECTION 4: First aid measures**

4.1 Description of first aid	
measures	
General information:	



After inhalation:	If breathed in, move person
Arter illialation.	
	immediately into fresh air. Consult
A.C. 1.	a physician
After skin contact:	Remove contaminated clothing.
	Wash with soap and water.
FCTD	Massage Calcium Gluconate Gel
F5111_	into the burnt area, continue this
	procedure until medical attention
	is available. Consult a physician.
After eye contact:	Immediately flush eyes with plenty
	of water or saline solution for at
	least 15 minutes. consult a
	physician.
After swallowing:	Immediately rinse mouth and
3	provide fresh air. Drink plenty of
	Milk of Magnesia or 6 effercescent
	soluble Calcium Tablets in water.
	Do not induce vomiting. Consult a
	physician.
4.2 Most important symptoms	can cause severe respiratory, skin,
and effects, both acute and	and eye damage, as well as
delayed:	gastrointestinal burns, while
30.0 <b>3</b> 0.00	delayed effects may include bone
	damage, kidney and liver
	dysfunction, and dental fluorosis
	from prolonged or repeated
	exposure.
4.3 Indication of any immediate	Treat symptomatically.
medical attention and special	Treat symptomatically.
treatment needed:	
treatment needed:	

## **SECTION 5: Firefighting measures**

5.1 Extinguishing media:	Extinguish with alcohol-resistant
	foam, carbon dioxide, dry powder
	or water fog.
5.2 Special hazards arising from	Hydrogen fluoride (HF). Oxides of:
the substance or mixture:	Fluorides
5.3 Advice for firefighters:	Wear fully protective suit, safety
	glasses and respiratory device .
	Cool tanks/drums with water
	spray/remove them into safety.
5.4 further information:	no data available



### **SECTION 6: Accidental release measures**

6.1 Personal precautions,	Use personal protective
protective equipment and	equipment.
emergency procedures:	Avoid breathing vapors, mist or
	gas. Ensure adequate ventilation.
	Remove all sources of ignition.
ECTD	Evacuate personnel to safe areas.
E9ID.	Beware of vapours accumulating
	to form explosive concentrations.
	Avoid dust accumulation. Seek
	medical attention.
6.2 Environmental precautions:	Do not enter this chemical into
	drains or into environment.
6.3 Methods and material for	Bund material with inert material,
containment and cleaning up:	carefully dilute with water spray to
	minimize fume emmision ahd heat
	generation. Carefully neutralize
	using excess of slaked lime or soda
	ash to form a slury. take up slurry
	into plastic containers and hold for
	disposal.

## **SECTION 7: Handling and storage**

7.1 Precautions for safe handling:	For use in are with adequate
	ventilation.
	Empty containers pose a fire risk,
	evaporate the residue under a
	fume hood. Ground all equipment
	containing material
	Do not use in confined spaces.
	Electrostatic discharge protection.
VOUD CHEMIC	Minimize dust generation and
IY U U K G F E WI G	accumulation.
	Avoid ingestion and inhalation.
7.2 Conditions for safe storage,	Store in original containers.
including any incompatibilities:	Keep containers securely sealed
Requirements to be met by	Keep container tightly closed in a
storerooms and receptacles:	dry and well-ventilated place.
	Containers which are opened must
	be carefully resealed and kept
	upright to prevent leakage.
7.3 Specific end uses:	no data available



## **SECTION 8: Exposure controls/personal protection**

8.1 Control parameters	
Additional information about	A system of local and general
design of technical facilities:	exhaust is recommended.
8.2 Exposure controls	
Appropriate engineering controls	Handle in accordance with good
ECTN	industrial hygiene and safety
	practice. Wash hands before
	breaks and at the end of workday.
Personal protective equipment:	Dust respirator, protective masks,
	wearing anti chemical gloves,
	rubber gloves, etc.
General protective and hygienic	Eyes, body and hand protection,
measures:	maintain indoor air unobstructed.
	Wear protective equipment.
	Respiratory protection: Required.
Protection of hands:	Handle with gloves. Gloves must be
	inspected prior to use. Use proper
	glove removal technique (without
	touching glove's outer surface) to
	avoid skin contact with this
	product. Dispose of contaminated
	gloves after use in accordance with
	applicable laws.
	Wash and dry hands.
	Eye protection: Required
Protection of Body:	Complete suit protecting against
	chemicals, Flame retardant
	antistatic protective clothing.

## SECTION 9: Physical and chemical properties

9.1 Information on basic physical and chemical properties General Information	
Appearance: Form:	Fuming liquid
Colour:	Colourless
Odour:	Strong
pH-value:	2
Melting point/Melting range:	-35°C
Boiling point/Boiling range:	19.5°C
Flammability (solid, gaseous):	Non- Flammable



Ignition temperature:	Not applicable
Decomposition temperature:	Not applicable
Self-igniting:	Not applicable
Flash point:	Not applicable
Danger of explosion:	Not applicable
Explosion limits: Lower:	Not applicable
Explosion limits: Upper:	Not applicable
Vapour pressure:	750 mmHg
Density at 20 °C:	1.15 g/cm <sup>3</sup> .
Relative density:	1.15
Vapour density:	16.5
<b>Evaporation rate:</b>	No data available
Solubility in / Miscibility with-	Completely miscible
·water at 20 °C:	
Partition coefficient:(n-	No data available
octanol/water)	
Viscosity:	1.2 cPs

## **SECTION 10: Stability and reactivity**

10.1 Reactivity	No data available
10.2 Chemical stability	This chemical is stable under
	storage conditions.
10.3 Possibility of hazardous	Under extreme heat, reactions can
reactions	happen releasing fluorides.
10.4 Conditions to avoid	Avoid excessive heat for prolonged
	periods of time
10.5 Incompatible materials	Inorganic sulphides. Inorganic
	cyanides. Organic cyanides
	(nitriles). Strong alkalis. Water,
	steam, water mixtures.
10.6 Hazardous decomposition	Toxic gases/vapours/fumes of:
products	Hydrogen fluoride (HF)

## **SECTION 11: Toxicological information**

11.1 Information on toxicological effects	
Acute Toxicity:	LD50 (Oral, Rat) : no data available LD50 (Dermal, Rabbit) : no data available LC50 (Inhalation Rat) : no data available



Skin corrosion/Irritation:	Causes burns
Serious eye damage/irritation:	Causes burns. Irritation, burning,
	lachrymation, blurred vision after
	liquid splash.
Respiratory damage/irritation:	Vapours irritate the respiratory
FCTD	system, and may cause coughing
F31U_	and difficulties in
	breathing.
Ingestion:	May cause chemical burns in
	mouth, oesophagus and stomach.
	Nausea, vomiting.
Germ cell mutagenicity:	No data available
Carcinogenicity:	No data available
Reproductive toxicity:	No data available
Specific target organ toxicity -	No data available
single exposure:	
Specific target organ toxicity -	No data available
repeated exposure:	
Aspiration hazard:	No data available
Signs and Symptoms of Exposure:	Refer section 2.3
11.2 Additional toxicological	
information	
Biodegradability:	not biodegradable.

## **SECTION 12: Ecological information**

12.1 Toxicity Aquatic toxicity:	LC50 (fish): 5.9-7.5 mg (96hr)
12.2 Persistence and	Not biodegradable.
degradability:	
12.3 Bioaccumulative potential:	No data available
12.4 Mobility in soil:	No data available
12.5 Other adverse effects:	No data available

## **SECTION 13: Disposal considerations**

13.1 Waste treatment methods	
Uncleaned packaging	dispose of in accordance with local
Recommendation:	hazardous waste regulations
Recommended cleansing agents:	Mild Soap and Water, Isopropyl
	Alcohol (IPA), sodium bicarbonate
	or sodium hydroxide.



#### **SECTION 14: Transport information**

14.1 UN-Number · ADR, ADN,	1790
IMDG, IATA:	
14.2 UN proper shipping name ·	HYDROFLUORIC ACID
ADR, ADN, IMDG, IATA:	
14.3 Transport hazard class(es) ·	8
ADR, ADN, IMDG, IATA :	1076
14.4 Packing group · ADR, IMDG,	2
IATA:	
14.5 Environmental hazards:	No data available
14.6 Special precautions for user:	Handle responsibly.

#### **SECTION 15: Regulatory information**

15.1 Safety, health and	Directive 2012/18/EU, under that
environmental	this substance is classified in listed
regulations/legislation specific	substance as corrosive.
for the substance or mixture	
Directive 2012/18/EU	
Named dangerous substances:	This substance is listed in the
	annex 1 to the directive.
15.2 Chemical safety assessment:	Chemical assessment has been
	carried out under <b>REACH</b>
	regulation.

#### **SECTION 16: Other information**

The data included was derived from international authoritative database and provided by the enterprise. Other information was based on the present state of our knowledge. We try to ensure the correctness of all information. However, due to the diversity of information sources and the limitations of our knowledge, this document is only for user's reference. Users should make their independent judgment of suitability of this information for their particular purposes. We do not assume responsibility for loss, damage or expense arising out of or in any way connected with the handling, storage, use or disposal of the product.